



Pergamon

Tetrahedron Letters 40 (1999) 7431-7433

TETRAHEDRON  
LETTERS

## A formal synthesis of ( $\pm$ ) parvifoline by manganese(III)-based oxidative arylation of ketones <sup>†</sup>

Dipal Ranjan Bhowmik and Ramanathapuram V. Venkateswaran \*

Department of Organic Chemistry, Indian Association for the Cultivation of Science, Jadavpur, Calcutta 700 032, India

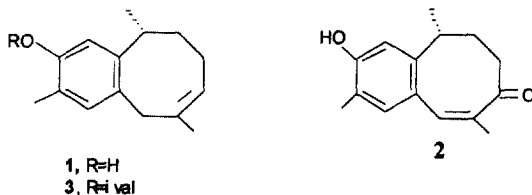
Received 29 June 1999; accepted 10 August 1999

### Abstract

A short, formal synthesis of the phenolic sesquiterpene ( $\pm$ ) parvifoline **1** is described involving manganese(III)-based oxidative arylation of the ketone **9** as the key step. © 1999 Published by Elsevier Science Ltd. All rights reserved.

**Keywords:** oxidative free-radical cyclization; Grob fragmentation; benzocyclooctene.

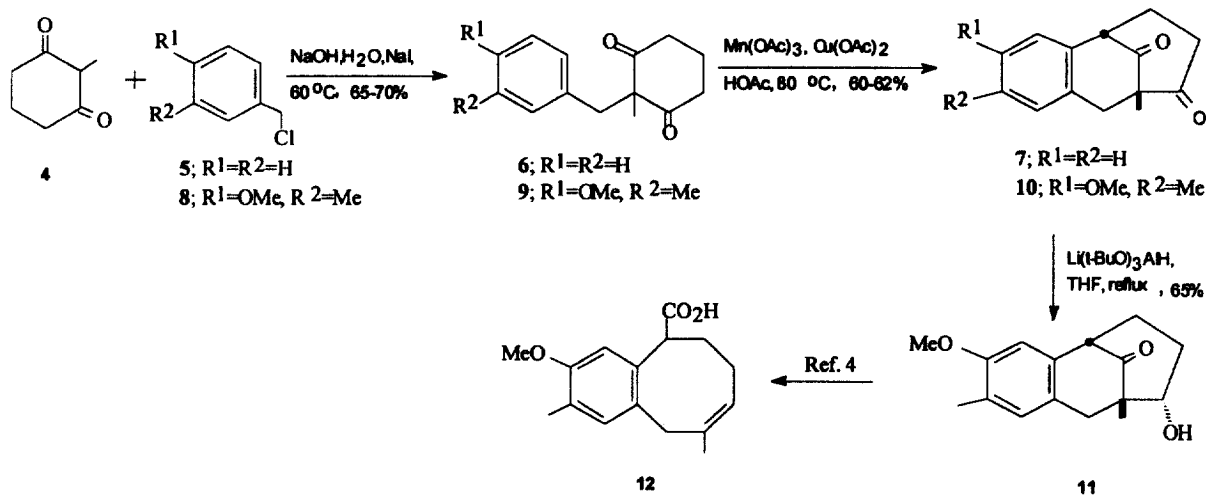
The sesquiterpenes parvifoline **1**, isoparvifolinone **2** and parvifoline isovalerate **3**, isolated from the species *Coreopsis* and *Perezia*, constitute the only examples of naturally occurring compounds containing a trimethyl benzocyclooctane structural unit.<sup>1</sup> Efforts leading to the synthesis of **1** and **2** have been reported.<sup>2-4</sup> We report a facile, formal synthesis of **1**, employing a manganese(III)-based oxidative cyclisation of a ketone onto an aromatic ring to furnish a benzobicyclo[3.3.1]nonane system followed by a Grob fragmentation to generate the benzocyclooctene core. Manganese(III)-based oxidative cyclisations and annulations involving relatively acidic compounds such as 1,3-diones, acetoacetates, malonates,  $\alpha$ -sulfinyl ketones etc. and unsaturated ketones have been extensively studied and their synthetic utility demonstrated.<sup>5</sup> However, there has been only one report on the manganese(III)-catalysed oxidative cyclisation of a ketone onto an aromatic ring leading to a new arylation process.<sup>6</sup> We have employed this protocol to prepare a benzobicyclo[3.3.1]nonane system which was fragmented to reveal the benzocyclooctene framework of **1**.



\* Corresponding author. Tel: (91) 33 473 4971; fax: (91) 33 473 2805; e-mail: ocrvv@mahendra.iacs.res.in

<sup>†</sup> Dedicated to the memory of Professor R. A. Raphael.

Alkylation<sup>7</sup> of 2-methylcyclohexane-1,3-dione **4** with benzyl chloride **5** afforded the C-alkylated product **6** in 70% yield (Scheme 1). Reaction of **6** in acetic acid with 2.5 equivalents of  $\text{Mn}(\text{OAc})_3$  and 1 equivalent of  $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  at 80°C for 5 h furnished the bridged bicyclo[3.3.1]nonandione **7** in 62% yield. The structural assignment was adequately supported by analytical and spectroscopic features.<sup>8</sup> In order to extend this methodology to a formal synthesis of **1**, dione **4** was alkylated with 3-methyl-4-methoxybenzyl chloride **8** to afford **9** in 65% yield. Oxidative cyclisation of **9** initiated by  $\text{Mn}(\text{OAc})_3$  as for **6** furnished in 60% yield the desired benzobicyclo[3.3.1]nonandione **10**. The two aromatic protons in **10** appeared as two singlets in the <sup>1</sup>H NMR spectrum establishing the regioselectivity of cyclisation. Regio- and stereoselective reduction of the least hindered carbonyl group in **10** was achieved by reaction with lithiumhydridotri *t*-butoxyaluminate<sup>9</sup> to produce the equatorial ketol **11** in 65% yield. Conversion to the corresponding mesylate followed by a Grob fragmentation as reported<sup>4</sup> afforded the benzocyclooctene carboxylic acid **12** in 80% overall yield. The spectroscopic features of the ketol **11** and the acid **12** and the melting point of **12** were in full agreement with those reported previously.<sup>4</sup> Since the acid **12** has been converted to parvifoline **1**<sup>4</sup> the present efforts constituted a formal synthesis of **1**. The synthesis thus demonstrates the versatility of manganese(III)-catalysed oxidative cyclisation of ketones onto aromatic rings to generate synthetically useful ring systems and the synthesis of an advanced intermediate is achieved in a short number of steps and good overall yield.



Scheme 1.

## Acknowledgements

We sincerely thank the Department of Science and Technology, New Delhi, Government of India for generous funding of this project.

## References

- Bohlmann, F.; Zdero, Ch. *Chem. Ber.* **1977**, *110*, 468; Joseph-Nathan, P.; Hernandez, J. D.; Roman, L. U.; Garcia, E.; Mendoza, V. *Phytochemistry* **1982**, *21*, 669; Joseph-Nathan, P.; Hernandez, J. D.; Roman, L. U.; Garcia, E.; Mendoza, V.; Mendoza, S. *Phytochemistry* **1982**, *21*, 1129; Garcia, E.; Mendoza, V.; Guzman, J. A. *J. Nat. Prod.* **1988**, *51*, 150.
- Grimm, E. L.; Levac, S.; Contu, M. L. *Tetrahedron Lett.* **1994**, *35*, 5369.

3. Villagomez-Ibarra, R.; Joseph-Nathan, P. *Tetrahedron Lett.* **1994**, *35*, 4771; Villagomez-Ibarra, R.; Alvarez-Cisneros, C.; Joseph-Nathan, P. *Tetrahedron* **1995**, *51*, 9285.
4. Covarrubias-Zuniga, A.; Cantu, F.; Maldonado, L. A. *J. Org. Chem.* **1998**, *63*, 2918.
5. For a recent review, see: Snider, B. B. *Chem. Rev.* **1996**, *96*, 339.
6. Cole, B. M.; Han, L.; Snider, B. B. *J. Org. Chem.* **1996**, *61*, 7832.
7. Schick, H.; Schwarz, H.; Finger, A. *Tetrahedron* **1982**, *38*, 1279.
8. All compounds reported herein gave analytical and spectral data in agreement with assigned structures. Selected spectral data for **7**: IR 1730, 1700  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz),  $\delta$  1.32 (s, 3H), 3.20 and 3.26 ( $\text{AB}_q$ , J 17.7 Hz, 2H) 3.84 (t, J 3.3 Hz, 1H); for **10**: IR 1735, 1699  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz)  $\delta$  1.30 (s, 3H), 2.17 (s, 3H), 3.11 and 3.21 ( $\text{AB}_q$ , J 17.1 Hz, 2H), 3.77 (t, J 3.3 Hz, 1H), 3.82 (s, 3H), 6.49 (s, 1H), 6.84 (s, 1H).
9. Colvin, E. W.; Martin, J.; Parker, W.; Raphael, R. A.; Shroot, B.; Doyle, M. J. *Chem. Soc., Perkin Trans. 1* **1972**, 860.